Synthesis of Some 1,3-Thiazole, 1,3,4-Thiadiazole, Pyrazolo[5,1-*c*]-1,2,4-triazine, and 1,2,4-Triazolo[5,1-*c*]-1,2,4-triazine Derivatives Based on the Thiazolo[3,2-*a*]benzimidazole Moiety

Nehal A. Hamdy¹, Hatem A. Abdel-Aziz^{1,*}, Ahmad M. Farag², and Issa M. I. Fakhr¹

¹ National Research Centre, Applied Organic Chemistry Department, Dokki, Cairo, Egypt

² Department of Chemistry, Faculty of Science, Cairo University, Giza, Egypt

Received May 9, 2007; accepted (revised) May 15, 2007; published online August 13, 2007 © Springer-Verlag 2007

Summary. 3-(3-Methylthiazolo[3,2-a]benzimidazol-2-yl)-3oxopropionitrile was synthesized by refluxing ethyl 3-methvlthiazolo[3,2-a]benzimidazole-2-carboxylate, acetonitrile, and sodium hydride. Treatment of 3-(3-methylthiazolo[3,2a]benzimidazol-2-yl)-3-oxopropionitrile with phenyl isothiocyanate, in the presence of KOH, furnished the corresponding potassium salt which was converted into thioacetanilide derivative upon neutralization. The thioacetanilide derivative reacts with α -chloroacetylacetone and ethyl α -chloroacetoacetate to give the 1,3-thiazole derivatives, while the reaction of the thioacetanilide derivative with hydrazonyl chlorides gave 1,3,4-thiadiazole derivatives. On the other hand, 3-(3methylthiazolo[3,2-a]benzimidazol-2-yl)-3-oxopropionitrile reacted with the diazonium salt of both 3-phenyl-5-amino-(1H)-pyrazole and 5-amino-1,2,4-(1H)-triazole to afford the corresponding hydrazones. The latter hydrazones underwent an intramolecular cyclization upon boiling in pyridine to give pyrazolo[5,1-c]-1,2,4-triazine and 1,2,4-triazolo[5,1-c]-1,2,4triazine derivatives. Moreover, the behavior of thiazolo[3,2a]benzimidazol-3(2H)-one towards phenyl isothiocyanate followed by the reaction with α -chloroketones or hydrazonyl chlorides was investigated. Some of the latter compounds exhibited moderate effects against some bacterial and fungal species.

Keywords. Heterocycles; Cyclizations; *Michael* addition; Hydrazonyl chlorides.

Introduction

Antibacterial [1-3], anti-inflammatory [4], antiulcer [5, 6], and antiviral [7, 8] effects have been shown

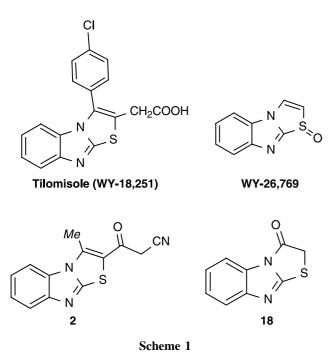
with various thiazolo[3,2-a]benzimidazole derivatives. Furthermore, certain thiazolo[3,2-a]benzimidazole derivatives, such as tilomisole (WY-18,251) was largely studied [9, 10] demonstrating their antiinflammatory [11] and immunomodulatory [12] activities. Recently, the gastric antisecretory activity of thiazolo[3,2-*a*]benzimidazol-1-oxide (WY-26,769) was reported [13]. Also, some thiazolo[3,2-a]benzimidazole derivatives were used for treatment of cancer [14], cerebral infarction [15], neurogenic pain [16], and bone diseases [17]. The above observations in connection with our research concerned with the synthesis of biologically active poly-substituted heterocycles [18-23] prompted us to synthesize some novel thiazolo[3,2-a]benzimidazole derivatives derived from 3-(3-methylthiazolo[3,2-a]benzimidazol-2-yl)-3-oxopropionitrile (2) and thiazolo[3,2a]benzimidazol-3(2H)-one (18) to evaluate their antimicrobial activities (Scheme 1).

Results and Discussion

Syntheses

The versatile, hitherto unreported 3-(3-methylthiazolo[3,2-a]benzimidazol-2-yl)-3-oxopropionitrile (2) was synthesized by refluxing equimolar quantities of 3-methylthiazolo[3,2-a]benzimidazole-2-carboxylic acid ethyl ester (1), acetonitrile and sodium hy-

^{*} Corresponding author. E-mail: hatem_741@yahoo.com

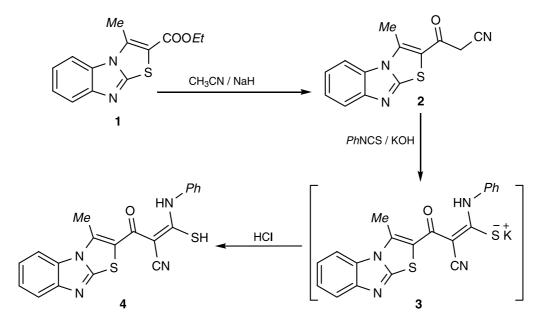


dride in dry benzene (Scheme 2). Compound **2** was characterized by its elemental analysis and spectral data. Thus, its IR spectrum revealed two characteristic absorption bands at 2225 and 1681 cm⁻¹ assignable to nitrile and carbonyl groups, whereas its ¹H NMR spectrum displayed a characteristic singlet signal at $\delta = 4.18$ due to active methylene protons in addition to aromatic protons in the region 7.21–8.01 ppm.

Next, treatment of compound 2 with phenyl isothiocyanate, in *DMF* and in the presence of KOH, at room temperature afforded the potassium salt 3which was converted into the thioacetanilide derivative 4 upon treatment with dilute hydrochloric acid (Scheme 2). The structure of compound 4was confirmed on the bases of its spectral data (*cf.* Experimental part).

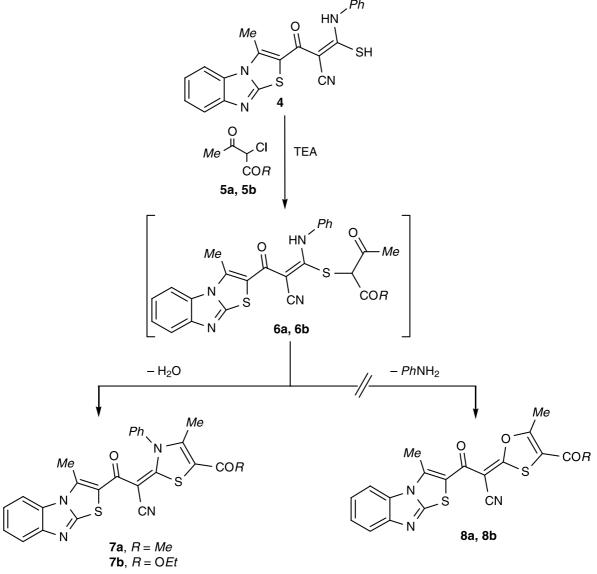
Compound 4 reacted with α -chloroacetylacetone (5a) and ethyl α -chloroacetoacetate (5b) in refluxing ethanol and in the presence of a catalytic amount of triethylamine resulted in the formation of 1,3-thiazole derivatives 7a and 7b [19] not the other possible oxathiole structures 8a and 8b [22] according to the spectral data of the isolated products (Scheme 3). For example, the IR spectra of the isolated products revealed, in each case, two bands due to two carbonyl groups in the region $1703-1638 \text{ cm}^{-1}$ and showed a nitrile absorption band in the region 2195- $2172 \,\mathrm{cm}^{-1}$. The mass spectra of the same compounds showed peaks corresponding to their molecular ions. The latter reaction of the thioacetanilide 4 with α chloroketones 5a and 5b proceeds, in each case, through loss of hydrogen chloride followed by elimination of a water molecule rather than elimination of aniline to afford 1.3-thiazole derivatives 7a and 7b not the other possible oxathiole structures 8a and 8b.

Furthermore, the reaction of thioacetanilide derivative 4 with hydrazonyl chlorides 9a-9c under the



Scheme 2

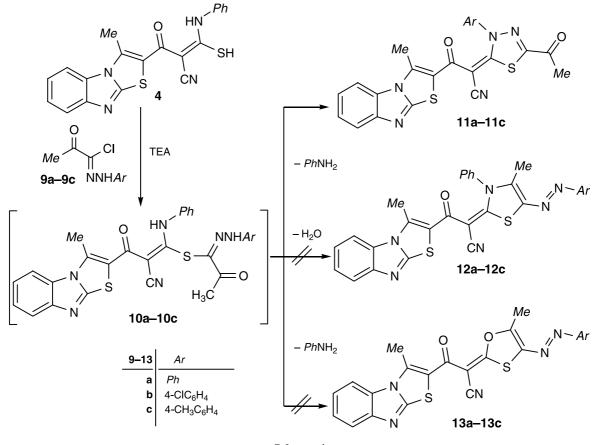






same reaction conditions afforded, in each case, only one isolable product. Scheme 4 depicts all the possible structures proposed for the reaction products. However, the elemental analyses and spectral data of the reaction products were compatible with the 1,3,4-thiadiazole structures **11a–11c**. For example, the IR spectra of the isolated products revealed, in each case, the appearance of two carbonyl absorption bands near 1695 and 1636 cm^{-1} in addition to a nitrile absorption band in the region $2205-2191 \text{ cm}^{-1}$. Their mass spectra revealed, in each case, a peak corresponding to the molecular ion. These results indicate that the reaction of the thioacetanilide **4** with hydrazonyl chlorides **9a–9c** proceeds in each case, *via* loss of hydrogen chloride to form the nonisolable intermediate **10a–10c** which cyclized by elimination of an aniline molecule to form the isolable 1,3,4-thiadiazole derivatives **11a–11c** [19, 22].

On the other hand, treatment of compound **2** with the diazonium salts of both 3-phenyl-5-amino-1*H*-pyrazole **14a** and 5-amino-1*H*-1,2,4-triazole **14b** [19, 24, 25] afforded hydrazones **15a** and **15b** (Scheme 5). The IR spectra of the isolated hydrazones exhibited, in each case, bands in the region $3325-3150 \text{ cm}^{-1}$ corresponding to 2NH functions and a nitrile absorption band in the region $2225-2218 \text{ cm}^{-1}$, in addition to a strong carbonyl band in the region $1660-1646 \text{ cm}^{-1}$.

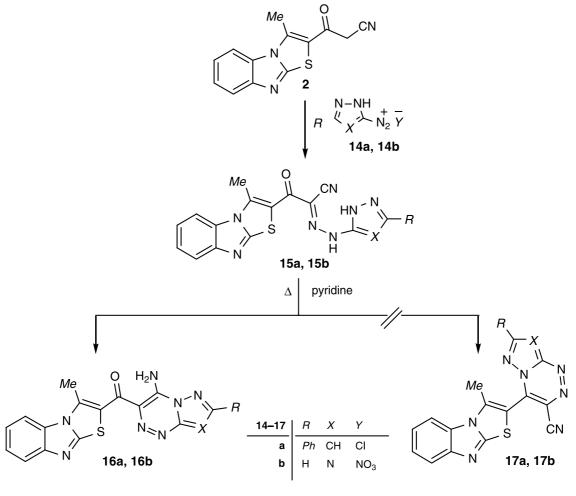


Scheme 4

Compounds **8a** and **8b** underwent an intramolecular cyclization upon boiling in pyridine *via Michael* type addition of the endocyclic NH of the hydrazones **8a** and **8b** to the triple bond of a nitrile function to afford the corresponding pyrazolo[5,1-*c*]-1,2,4-triazine and 1,2,4-triazolo[5,1-*c*]-1,2,4-triazine derivatives **16a** and **16b** (Scheme 5). Their IR spectra revealed, in each case, the lack of band corresponding to nitrile absorption and revealed the presence of amino and carbonyl functions, whereas, their ¹H NMR showed, in each case, the appearance of a D₂O exchangeable signal in the region 8.55–9.06 ppm due to the amino function (*cf.* Experimental part).

The behavior of thiazolo[3,2-*a*]benzimidazol-3(2*H*)-one (**21**) towards phenyl isothiocyanate followed by the reaction with α -chloroacetylacetone (**5a**), ethyl α -chloroacetoacetate (**5b**), or hydrazonyl chlorides **9a–9c** was investigated. Thus, treatment of thiazolo[3,2-*a*]benzimidazol-3(2*H*)-one (**18**) with phenyl isothiocyanate, in *DMF* and in the presence of KOH, at ambient temperature furnished the nonisolable potassium salt **19** which reacts *in situ* with α -chloroacetylacetone (**5a**) and ethyl α -chloroacetoacetate (**5b**) to give 1,3-thiazole derivatives **20a** and **20b** (Scheme 6). In a similar manner, hydrazonyl chlorides **9a–9e** reacted with the non-isolable potassium salt **19** under the same reaction conditions, to afford 1,3,4-thiadiazole derivatives **21a–21e**. The structures of the latter compounds were confirmed by their IR, ¹H NMR, and mass spectra (*cf.* Experimental part). The latter reactions of the non-isolable potassium salt **19** with α -chloroketones **5a** and **5b** or hydrazonyl chlorides **9a–9c** proceeds in the same sequence of these reactions depicted in Schemes 3 and 4, respectively, except elimination of potassium chloride instead of hydrogen chloride [19, 26].

Compound **18** couples smoothly with 3-phenyl-1*H*-pyrazole-5-diazonium chloride (**14a**) and 1*H*-1,2,4-triazole-5-diazonium nitrate **14b** to afford the corresponding hydrazones **22a** and **22b** (Scheme 6). The structures of the latter hydrazones were deduced by their elemental analyses and spectral data. For example, the IR spectrum of the hydrazone deriva-

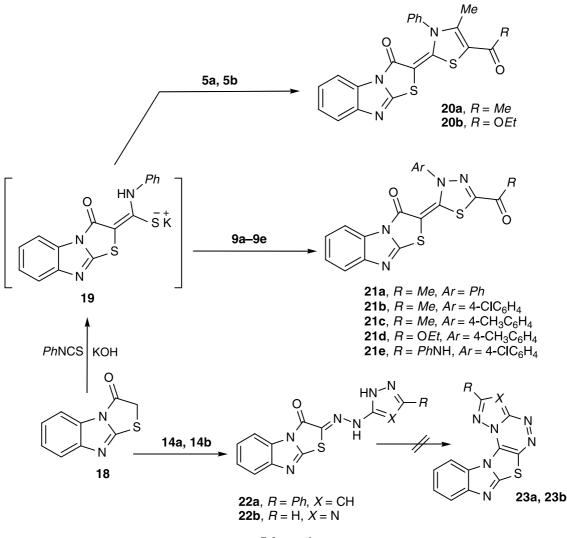




tive **22a** showed bands at 3208 and 3112 cm^{-1} due to two NH groups and exhibited a strong absorption band at 1701 cm⁻¹ corresponding to a carbonyl group while its ¹H NMR revealed two D₂O exchangeable signals of two NH functions at $\delta = 8.65$, 10.44 ppm. All attempts to cyclize the hydrazones **22a** and **22b** to their corresponding fused heterocyclic ring systems **23a** and **23b** were unsuccessful.

Antimicrobial Activity

The antibacterial and antifungal activities were carried out at the Regional Center for Mycology and Biotechnology at Al-Azhar University, Cairo, Egypt. Some of the newly synthesized compounds were screened for their antimicrobial activity using the diffusion agar technique [27], where ten compounds were tested against four fungal species namely *Aspergillus fumigatus (A.f.), Penicillium italicum* (P.i.), Syncephalastrum racemosum (S.r.), and Candida albicans (C.a.) as well as against four bacterial species namely Staphylococcus aureus (S.a.), Pseudomonas aeruginosa (P.a.), Bacillus subtilis (B.s.), and Escherichia coli (E.c.) for their antimicrobial activity using 5 mg/cm^3 of each compound in DMF. Inhibition zone diameter (IZD) in cm was taken as the criterion for antimicrobial activity. The antimicrobial activity was assayed biologically using a spore suspension of the fungal species (1 cm^3) of sterile water containing approximately 10⁸ conidia) or spreading bacterial suspension over a solidified malt agar medium. The layer was allowed to set for 30 min. A solution of each of the tested compounds (5 mg/cm^3) was placed onto sterile 5 mm filter paper discs and allowed to dry, then the discs were placed onto the center of the malt agar plate and incubated at the optimum incubation temperature, $28 \pm 2^{\circ}$ C. A clear zone around the disc was taken as an indication



Scheme 6

Compd. ^a	Inhibitions ^b Fungal species				Inhibitions ^b Bacterial species			
	7b	0	0	+	+	0	0	+
11c	0	+	0	+	0	+	0	0
16a	0	0	+	+	0	+	0	0
20b	0	+	++	+	+	++	++	+
21c	+	0	+	+	0	+	0	+
22b	+	+	0	+	0	+	++	0
Stand. ^c	+++	+++	+++	++	++	+++	+++	++

Table 1. Antimicrobial activities of the tested compounds

^a The biological activity of the other compounds 7a, 11a, 16b, and 21d revealed no inhibition activities against all species

^b Inhibition value = 0.1-0.5 cm beyond control = +, 0.6-1.0 cm = ++, 1.1-1.5 cm = +++ and 0 = not detected

^c The fungicide *Terbinafin* and the bactericide *Chloramphenicol* were used as standards

of the inhibition of the test organism growth. The size of the clear zone is proportional to the inhibitory action of the compound under investigation. The fungicide *Terbinafin* and the bactericide *Chloramphenicol* were used as references to evaluate the potency of the tested compounds under the same conditions. Measurements were considered after 72 h for fungi and 24 h for bacteria. The results are contained in Table 1.

The results of antimicrobial activity showed that compound **20b** exhibited moderate effects against *Syncephalastrum racemosum* (*S.r.*), *Pseudomonas aeruginosa* (*P.a.*) and *Bacillus subtilis* (*B.s.*). Also, the hydrazone derivative **22b** showed a moderate effect against *Bacillus subtilis* (*B.s.*). Most of tested compounds exhibited slightly effects against *Candida albicans* (*C.a.*) and *Pseudomonas aeruginosa* (*P.a.*). However, the activities of the tested compounds are less than those of the standard agents used (Table 1).

Experimental

Melting points were measured with a Gallenkamp apparatus. IR spectra were recorded on a Shimadzu FT-IR 8101 PC infrared spectrophotometer. The ¹H NMR spectra were determined in *DMSO*-d₆ at 300 MHz on a Varian Mercury VX 300 NMR spectrometer using *TMS* as an internal standard. Mass spectra were measured on a GCMS-QP1000 EX spectrometer at 70 eV. Elemental analyses (C, H, N, and S) were performed using the Elmentar varu EL Germany Instrument. Their results agreed favourably with the calculated values. 3-Methylthiazolo[3,2-*a*]benzimidazole-2-carboxylic acid ethyl ester (1) [28], thiazolo[3,2-*a*]benzimidazol-3(2*H*)-one (18) [29], hydrazonyl chlorides **9a–9c** [30], **9d** [31], and **9e** [32] were prepared according to the reported methods.

*3-(3-Methylthiazolo[3,2-a]benzimidazol-2-yl)-3*oxopropionitrile (**2**, C₁₃H₉N₃OS)

To 26.0 g 3-methylthiazolo[3,2-*a*]benzimidazole-2-carboxylic acid ethyl ester (1) (100 mmol) dissolved in 250 cm³ dry benzene, 10 cm³ *DMF*, 4.1 g acetonitrile (100 mmol), and 4.8 g sodium hydride (60%) were added. The reaction mixture was refluxed for 4 h, then allowed to cool. The solid that precipitated was collected by filtration, washed with ether, and dried. The solid product was dissolved in water and the resulting solution was acidified with concentrated HCl until it became neutral (*pH* 7). The precipitated solid was collected by filtration, washed with water, and dried. Recrystallization of the crude product from *Et*OH/*DMF* gave 3-(3-methylthiazolo[3,2-*a*]benzimidazol-2-yl)-3-oxopropionitrile (**2**) as white powder in 58% yield, mp 210–212°C; IR (KBr): $\bar{\nu} = 2225$ (C \equiv N), 1681 (C=O), 1612 (C=N) cm⁻¹; ¹H NMR (300 MHz, *DMSO*-d₆): $\delta = 3.04$ (s, 3H, CH₃), 4.18 (s, 2H, CH₂), 7.21–

8.01 (m, 4H, ArH) ppm; MS (70 eV): m/z = 256 (M⁺ + 1), 255 (M⁺), 189 (100%), 118, 67.

3-Mercapto-2-(3-methylthiazolo[3,2-a]benzimidazol-2-oyl)-3-phenylamino-acrylonitrile (4, $C_{20}H_{14}N_4OS_2$)

To a stirred solution of 0.56 g KOH (10 mmol) in 20 cm³ *DMF*, 2.55 g propionitrile derivative **2** (10 mmol) was added. After stirring for 30 min 1.35 g phenyl isothiocyanate (10 mmol) was added to the resulting mixture. Stirring was continued for 6 h, then poured over crushed ice containing HCl. The solid product so formed was filtered off, washed with water, dried, and finally crystallization from *Et*OH/*DMF* afforded compound **4** as yellow solid in 86% yield, mp 221–223°C; IR (KBr): $\bar{\nu} = 3206$ (NH), 2183 (C \equiv N), 1635 (C=O), 1558 (C=N) cm⁻¹; ¹H NMR (300 MHz, *DMSO*-d₆): $\delta = 3.01$ (s, 3H, CH₃), 7.26–8.12 (m, 9H, ArH), 10.70 (s, 1H, NH, D₂O exchangeable), 12.50 (s, 1H, SH, D₂O exchangeable) ppm; MS (70 eV): *m*/*z* = 391 (M⁺ + 1), 390 (M⁺), 224 (100%), 188, 134, 77.

Reaction of Thioacetanilide **4** with α -Chloroketones **5a**, **5b**. General Procedure

To 0.39 g 4 (1 mmol) dissolved in 20 cm³ ethanol, the appropriate α -chloroacetylacetone (5a) or ethyl α -chloroacetoacetate (5b) (1 mmol), and 0.5 cm³ triethylamine were added. The mixture was refluxed for 8 h, then allowed to cool. The formed solid was filtered off, washed with ethanol, and recrystallized from *DMF*/H₂O to afford the corresponding 1,3-thiazole derivatives 7a and 7b.

$\begin{array}{l} 2\mbox{-}[5\mbox{-}Acetyl\mbox{-}4\mbox{-}methyl\mbox{-}3\mbox{-}phenyl\mbox{-}3\mbox{-}H\mbox{-}thiazol\mbox{-}2\mbox{-}yl\mbox{-}a\mbox{-}a\mbox{-}phenyl\mbox{-}3\mbox{-}c\mbox{-}x\mbox{-}phenyl\mbox{-}a\mbox{-}b\mbox{-}a\mbox{-}a\mbox{-}a\mbox{-}b\mbox{-}a\mbox{-$

This compound was produced as yellow crystals, yield 54%, mp 271–273°C; IR (KBr): $\bar{\nu} = 2195$ (C \equiv N), 1654, 1638 (2C=O), 1587 (C=N) cm⁻¹; ¹H NMR (300 MHz, *DMSO*-d₆): $\delta = 2.24$ (s, 3H, CH₃), 2.60 (s, 3H, CH₃), 2.81 (s, 3H, CH₃), 2.81 (s, 3H, CH₃), 2.81 (s, 3H, CH₃), 2.81 (s, 3H, CH₃), 2.40 (s, 3H, CH₃), 2.81 (s, 3H, CH₃), 7.24–8.12 (m, 9H, ArH) ppm; MS (70 eV): m/z = 471 (M⁺ + 1), 470 (M⁺), 380 (100%), 225, 153, 118, 98.

2-[1-Cyano-2-(3-methylthiazolo[3,2-a]benzimidazol-2-yl)-2-oxoethylidene]-4-methyl-3-phenyl-2,3-dihydrothiazole-5-carboxylic acid ethyl ester (**7b**, C₂₆H₂₀N₄O₃S₂)

This compound was produced as yellow crystals, yield 57%, mp 241–242°C; IR (KBr): $\bar{\nu} = 2172$ (C \equiv N), 1703, 1655 (2C=O), 1598 (C=N) cm⁻¹; ¹H NMR (300 MHz, *DMSO*-d₆): $\delta = 2.02$ (t, 3H, J = 7.26 Hz, CH₃), 2.98 (s, 3H, CH₃), 4.13 (q, 2H, J = 7.26 Hz, CH₂), 7.22–8.09 (m, 9H, ArH) ppm; MS (70 eV): m/z = 501 (M⁺ + 1), 500 (M⁺), 225 (100%), 188, 132, 68.

Reaction of Thioacetanilide **4** with Hydrazonyl Chlorides **9a–9c**

This reaction was carried out by the same procedure described above using the appropriate hydrazonyl chlorides **9a–9c** instead of the α -chloroketones **5a** and **5b**. The solid product was recrystallized from *DMF*/H₂O to afford the corresponding 1,3,4-thiadizole derivatives **11a** and **11b**.

2-[5-Acetyl-3-phenyl-3H-1,3,4-thiadiazol-2-ylidene]-3-(3methylthiazolo[3,2-a]benzimidazol-2-yl)-3-oxopropionitrile (**11a**, C₂₃H₁₅N₅O₂S₂)

Yellow crystals, yield 76%, mp 277–279°C; IR (KBr): $\bar{\nu} = 2205$ (C=N), 1695, 1636 (2C=O), 1580 (C=N) cm⁻¹; ¹H NMR (300 MHz, *DMSO*-d₆): $\delta = 2.43$ (s, 3H, CH₃), 2.89 (s, 3H, CH₃), 7.13–7.95 (m, 9H, ArH) ppm; MS (70 eV): m/z = 458 (M⁺ + 1), 457 (M⁺, 100%), 241 (7.4%), 77.

2-[5-Acetyl-3-(4-chlorophenyl)-3H-1,3,4-thiadiazol-2ylidene]-3-(3-methylthiazolo[3,2-a]benzimidazol-2-yl)-3oxopropionitrile (**11b**, C₂₃H₁₄ClN₅O₂S₂)

Yellow crystals, yield 72%, mp >300°C; IR (KBr): $\bar{\nu} = 2191$ (C \equiv N), 1686, 1640 (2C=O), 1573 (C=N) cm⁻¹; ¹H NMR (300 MHz, *DMSO*-d₆): $\delta = 2.49$ (s, 3H, CH₃), 2.92 (s, 3H, CH₃), 7.27–7.06 (m, 8H, ArH) ppm; MS (70 eV): m/z = 493(M⁺ + 2), 492 (M⁺ + 1), 491 (M⁺), 342 (100%), 243, 187, 113.

$\begin{array}{l} 2\mbox{-}[5\mbox{-}Acetyl\mbox{-}3\mbox{-}(4\mbox{-}tolyl\mbox{)\mbox{-}}3\mbox{-}1\mbox{,}3\mbox{-}4\mbox{-}tolyl\mbox{)\mbox{-}}3\mbox{-}2\mbox{-}yl\mbox{)\mbox{-}}3\mbox{-}copropionitrile \\ (11c, C_{24}H_{17}N_5O_2S_2) \end{array}$

Yellow needles, yield 78%, mp >300°C; IR (KBr): $\bar{\nu} = 2199$ (C=N), 1679, 1646 (2C=O), 1584 (C=N) cm⁻¹; ¹H NMR (300 MHz, *DMSO*-d_6): $\delta = 2.23$ (s, 3H, CH₃), 2.45 (s, 3H, CH₃), 2.61 (s, 3H, CH₃), 7.07–7.73 (m, 8H, ArH) ppm; MS (70 eV): m/z = 472 (M⁺ + 1), 471 (M⁺, 100%), 346, 294, 146, 94, 51.

Reaction of 3-(3-Methylthiazolo[3,2-a]benzimidazol-2-yl)-3-oxopropionitrile (2) with 3-Phenyl-1H-pyrazole-5diazonium Chloride (14a) and 1H-1,2,4-Triazole-5diazonium Nitrate (14b). General Procedure

To 2.55 g 2 (10 mmol) dissolved in 30 cm³ pyridine, the appropriate diazonium salt 14a or 14b (10 mmol) was added. The addition was carried out portion-wise with stirring at $0-5^{\circ}$ C over a period of 30 min. After complete addition, the reaction mixture was stirred for further 4 h, then kept in an icechest for 12 h, and finally diluted with H₂O. The precipitated solid was collected by filtration, washed with water and dried. Recrystallization from *Et*OH/*DMF* afforded the corresponding hydrazone derivatives 15a and 15b.

3-(3-Methylthiazolo[3,2-a]benzimidazol-2-yl)-3-oxo-2-[(5-phenyl-2H-pyrazol-3-yl)hydrazono]propionitrile (15a, C₂₂H₁₅N₇OS)

Yellow solid, yield 61%, mp >300°C; IR (KBr): $\bar{\nu}$ = 3325, 3163 (2NH), 2218 (C=N), 1660 (C=O), 1620 (C=N) cm⁻¹; MS (70 eV): m/z = 426 (M⁺ + 1), 425 (M⁺, 100%), 302, 291, 77.

3-(3-Methylthiazolo[3,2-a]benzimidazol-2-yl)-3-oxo-2-[2H-1,2,4-triazol-3-yl)hydrazono] propionitrile

(15b, C₁₅H₁₀N₈OS)

Yellow solid, yield 57%, mp >300°C; IR (KBr): $\bar{\nu}$ = 3282, 3150 (2NH), 2225 (C=N), 1646 (C=O), 1616 (C=N) cm⁻¹;

MS (70 eV): m/z = 352 (M⁺ + 2), 351 (M⁺ + 1), 350 (M⁺, 100%), 189, 91.

Intramolecular Cyclization of the Heterocyclic Hydrazones **15a** and **15b**. General Procedure

A solution of the appropriate hydrazone **15a** and **15b** (1 mmol) in 15 cm³ pyridine was refluxed for 6 h, then left to cool. The solid that formed was filtered off, washed with ethanol, and dried. Recrystallization from DMF/H_2O afforded the corresponding fused heterocyclic **16a** and **16b**.

2-(4-Amino-7-phenylpyrazolo[5,1-c]-1,2,4-triazin-3-oyl)-3methylthiazolo[3,2-a]benzimidazole (**16a**, C₂₂H₁₅N₇OS) Brown solid, yield 67%, mp >300°C; IR (KBr): $\bar{\nu}$ = 3130, 3235 (NH₂), 1624 (C=O), 1558 (C=N) cm⁻¹; ¹H NMR (300 MHz, *DMSO*-d₆): δ = 3.10 (s, 3H, CH₃), 6.57 (s, 1H, pyrazole), 7.21–8.03 (m, 9H, ArH), 9.06 (br, s, 2H, NH₂, D₂O exchangeable) ppm; MS (70 eV): m/z = 427 (M⁺ + 2), 426 (M⁺ + 1), 425 (M⁺, 100%), 324, 291, 187, 105, 77.

2-(4-Amino-1,2,4-triazolo[5,1-c]-1,2,4-triazin-3-oyl)-3methylthiazolo[3,2-a]benzimidazole (**16b**, C₁₅H₁₀N₈OS) Pale brown solid, yield 65%, mp >300°C; IR (KBr): $\bar{\nu}$ = 3233, 3105 (NH₂), 1612 (C=O), 1535 (C=N) cm⁻¹; ¹H NMR (300 MHz, *DMSO*-d₆): δ = 3.14 (s, 3H, CH₃), 7.24–8.02 (m, 4H, ArH), 8.28 (s, 1H, triazole), 8.55 (br, s, 2H, NH₂, D₂O exchangeable) ppm; MS (70 eV): m/z = 352 (M⁺ + 2), 351 (M⁺ + 1), 350 (M⁺), 284, 188, 118 (100%), 105, 91, 72.

Reaction of Thiazolo[3,2-a]benzimidazol-3(2H)-one (18) with Phenyl isothiocyanate

To 0.06 g KOH (1 mmol) in 20 cm³ *DMF*, thiazolo[3,2-*a*]benzimidazol-3(2*H*)-one (**18**) (1 mmol) was added. After stirring for 30 min 0.135 g phenyl isothiocyanate (1 mmol) were added to the resulting mixture. Stirring was continued for 6 h, then the appropriate α -chloroketones **5a** and **5b** or hydrazonyl chlorides **9a–9e** (1 mmol) was added portion-wise over a period of 30 min. After the addition was complete, the reaction mixture was stirred for 12 h. The yellowish colored precipitate was filtered off, washed with H₂O, and dried. Crystallization from *Et*OH/*DMF* afforded the corresponding 1,3-thiazole derivatives **20a** and **20b**, and 1,3,4-thiadiazole derivatives **21a–21e**.

2-[5-Acetyl-4-methyl-3-phenyl-3H-thiazol-2-ylidene]-3-oxothiazolo[3,2-a]benzimidazole (**20a**, C₂₁H₁₅N₃O₂S₂)

Yellow crystals, yield 55%, mp >300°C; IR (KBr): $\bar{\nu} = 1666$, 1638 (2C=O), 1612 (C=N) cm⁻¹; ¹H NMR (300 MHz, *DMSO*-d₆): $\delta = 2.22$ (s, 3H, CH₃), 3.16 (s, 3H, CH₃), 7.23–8.01 (m, 9H, ArH) ppm; MS (70 eV): m/z = 407 (M⁺ + 2), 406 (M⁺ + 1), 405 (M⁺, 100%), 365, 244, 188, 164, 107, 77.

$\label{eq:alpha} \begin{array}{l} \mbox{4-Methyl-2-(3-oxothiazolo[3,2-a]benzimidazol-2-ylidene)-3-phenyl-2,3-dihydrothiazole-5-carboxylic acid ethyl ester} \\ \mbox{(20b, $C_{22}H_{17}N_3O_3S_2$)} \end{array}$

Yellow needles, yield 59%, mp >300°C; IR (KBr): $\bar{\nu} = 1684, 1649 (2C=O), 1607 (C=N) \text{ cm}^{-1}; {}^{1}\text{H NMR} (300 \text{ MHz},$

DMSO-d₆): $\delta = 1.32$ (t, 3H, J = 7.25 Hz, CH₃), 2.18 (s, 3H, CH₃), 4.31 (q, 2H, J = 7.25 Hz, CH₂), 7.26–7.99 (m, 9H, ArH) ppm; MS (70 eV): m/z = 437 (M⁺ + 2), 436 (M⁺ + 1), 435 (M⁺, 100%), 407, 363, 258, 231, 187, 77.

2-(5-Acetyl-3-phenyl-3H-1,3,4-thiadiazol-2-ylidene)-3-oxo-

thiazolo[3,2-a]benzimidazole (**21a**, C₁₉H₁₂N₄O₂S₂) Yellow needles, yield 56%, mp 290–292°C; IR (KBr): $\bar{\nu} = 1689$, 1663 (2C=O), 1609 (C=N) cm⁻¹; ¹H NMR (300 MHz, *DMSO*-d₆): $\delta = 2.34$ (s, 3H, CH₃), 7.28–8.00 (m, 9H, ArH) ppm; MS (70 eV): m/z = 394 (M⁺ + 2), 393 (M⁺ + 1), 392 (M⁺, 100%), 291, 243, 205, 143, 103, 77.

2-[5-Acetyl-3-(4-chlorophenyl)-3H-1,3,4-thiadiazol-2ylidene]-3-oxothiazolo[3,2-a]benzimidazole

 $(21b, C_{19}H_{11}CIN_4O_2S_2)$

Yellow needles, yield 60%, mp 227–229°C; IR (KBr): $\bar{\nu} =$ 1693, 1655 (2C=O), 1593 (C=N) cm⁻¹; ¹H NMR (300 MHz, *DMSO*-d₆): $\delta = 2.34$ (s, 3H, CH₃), 7.27–7.81 (m, 8H, ArH) ppm; MS (70 eV): m/z = 428 (M⁺ + 2), 427 (M⁺ + 1), 426 (M⁺, 100%), 316, 283, 225, 146, 109.

2-[5-Acetyl-3-(4-tolyl)-3H-1,3,4-thiadiazol-2-ylidene]-3-oxothiazolo[3,2-a]benzimidazole (**21c**, C₂₀H₁₄N₄O₂S₂)

Yellow needles, yield 53%, mp 280–282°C; IR (KBr): $\bar{\nu} =$ 1703, 1648 (2C=O), 1602 (C=N) cm⁻¹; ¹H NMR (300 MHz, *DMSO*-d₆): $\delta = 2.30$ (s, 3H, CH₃), 2.33 (s, 3H, CH₃), 7.24–7.78 (m, 8H, ArH) ppm; MS (70 eV): m/z = 407 (M⁺ + 1), 406 (M⁺, 100%), 299, 196, 109.

5-(3-Oxothiazolo[3,2-a]benzimidazol-2-ylidene)-4-(4-tolyl)-4,5-dihydro-1,3,4-thiadiazole-2-carboxylic acid ethyl ester (**21d**, C₂₁H₁₆N₄O₃S₂)

Yellow crystals, yield 64%, mp 220–222°C; IR (KBr): $\bar{\nu} =$ 1741, 1683 (2C=O), 1593 (C=N) cm⁻¹; ¹H NMR (300 MHz, *DMSO*-d₆): $\delta =$ 1.26 (t, 3H, J = 7.26 Hz, CH₃), 2.33 (s, 3H, CH₃), 4.22 (q, 2H, J = 7.26 Hz, CH₂), 7.19–7.77 (m, 8H, ArH) ppm; MS (70 eV): m/z = 437 (M⁺ + 1), 436 (M⁺), 377 (100%), 324, 164, 105.

2-(*Phenylcarboxamido*)-4-(4-chlorophenyl)-5-(3-oxothiazolo[3,2-a]benzimidazol-2-ylidene)-4,5-dihydro-1,3,4thiadiazole (**21e**, C₂₄H₁₄ClN₅O₂S₂)

Pale yellow solid, yield 62% mp >300°C; IR (KBr): $\bar{\nu}$ = 3165 (NH), 1692, 1638 (C=O), 1577 (C=N) cm⁻¹; ¹H NMR (300 MHz, *DMSO*-d₆): δ = 7.21–7.94 (m, 14H, ArH), 8.64 (br. s, 1H, NH, D₂O exchangeable) ppm; MS (70 eV): m/z = 507 (M⁺ + 4), 505 (M⁺ + 2), 504 (M⁺ + 1), 503 (M⁺), 355 (100%), 218, 151, 86.

Reaction of Thiazolo[3,2-a]benzimidazol-3(2H)-one (18) with Diazonium Salts of Heterocyclic Amines 14a and 14b. General Procedure

This procedure is similar to the method that was used in synthesis of compounds **15a** and **15b** above using thiazolone derivative **18** instead of compound **2**. The corresponding hydrazones **22a** and **22b** were crystallized from DMF/H_2O .

2-[(5-Phenyl-2H-pyrazol-3-yl)hydrazono]-3-oxothiazolo-[3,2-a]benzimidazole (**22a**, C₁₈H₁₂N₆OS)

Yellow solid, yield 64%, mp >300°C; IR (KBr): $\bar{\nu} = 3208$,

3112 (2NH), 1701 (C=O), 1616 (C=N) cm⁻¹; ¹H NMR (300 MHz, *DMSO*-d₆): δ = 5.65 (s, 1H, pyrazole), 7.31–7.93 (m, 9H, ArH), 8.65 (s, 1H, NH, D₂O exchangeable), 10.44 (s, 1H, NH, D₂O exchangeable) ppm; MS (70 eV): m/z = 363 (M⁺ + 3), 362 (M⁺ + 2), 361 (M⁺ + 1), 360 (M⁺, 100%), 318, 176, 160, 148, 77.

2-[(2H-1,2,4-Triazol-3-yl)hydrazono]-3-oxothiazolo[3,2-a]benzimidazole (**22b**, C₁₁H₇N₇OS)

Yellow solid, yield 59%, mp >300°C; IR (KBr): $\bar{\nu}$ = 3233, 3105 (2NH), 1705 (C=O), 1620 (C=N) cm⁻¹; ¹H NMR (300 MHz, *DMSO*-d₆): δ = 7.15 (s, 1H, NH, D₂O exchangeable), 7.31–7.94 (m, 4H, ArH), 8.27 (s, 1H, triazole), 8.80 (s, 1H, NH, D₂O exchangeable) ppm; MS (70 eV): m/z = 287 (M⁺ + 2), 286 (M⁺ + 1), 285 (M⁺, 100%), 257, 176, 160, 148, 90.

References

- Venkatesan AM, Mansour TS, Abe T, Mihira A, Agarwal A, Ushirogochi H, Gu Y, Tamai S, Sum F (2002) PCT Int Appl WO 03 93,280; (2003) Chem Abstr 139: 381302u
- [2] Oh C, Ham Y, Hong S, Cho J (1995) Arch Pharm 328: 289
- [3] Allanson NM, Leslie BW, Thomson S (2001) Brit UK Pat Appl GB 2,376,944; (2003) Chem Abstr 138: 55966t
- [4] Bender PE, Hill D, Offen PH, Razgaitis K, Lavanchy P, Stringer OD, Sutton BM, Giswold DE, Dimertino M (1985) J Med Chem 28: 1169
- [5] Yoshida A, Oda K, Tabata K (1991) Jpn Kokai Tokkyo Koho JP 03 14,566; (1991) Chem Abstr 115: 71660z
- [6] Crossley R (1989) U.S. US 4,873,237; (1990) Chem Abstr 112: 198401d
- [7] Moormann AE, Becker DP, Flynn DL, Li H, Villamil CI (1999) U.S. US 5,945,425; (1999) Chem Abstr 131: 184948t
- [8] Moormann AE, Becker DP, Flynn DL, Li H, Vilamil CI (1995) PCT Int Appl WO 95,29,897; (1996) Chem Abstr 124: 202255b
- [9] Rainsford KD (1989) J Pharm Pharmacol 41(2): 112
- [10] Liveridge GG, Conzentino P, Cundy KC, Sarpotdar PP (1993) PCT Int Appl WO 93 25,190; (1994) Chem Abstr 120: 144167d
- [11] Eickhoff WM, Engers DA, Mueller KR (1996) U.S. US 96 24,336; (1996) Chem Abstr 125: 41845x
- [12] Gilman SC, Carlson RP, Daniels JF, Datko L, Berner PR, Chang J, Lewis AJ (1987) Int J Immunopharmacol 9: 9
- [13] Dijoseph JF, Palumbo GJ, Crossley R, Santili AA, Nielsen ST (1991) Drug Dev Res 23: 57; (1991) Chem Abstr 115: 64464e
- [14] Mckee TD, Suto RK (2002) PCT Int Appl WO 03 73,999; (2003) Chem Abstr 139: 240337x
- [15] Itahana H, Okada S, Ohara A, Negoro K, Nozawa S, Kamikubo T, Sakamoto S (2002) Jpn Kokai Tokkyo Koho JP 105,085; (2002) Chem Abstr 136: 294827p

- [16] Okada M, Nagakura Y, Kiso T, Toya T, Hayashibe S(2001) PCT Int Appl WO 01, 08,705; Chem Abstr 134: 141763y
- [17] Oku T, Kawai Y, Yatabe T, Sato S, Yamazaki H, Kayakiri N, Yoshihara K (1997) PCT Int Appl WO 97 10,219; Chem Abstr 126: 293352m
- [18] Abdel-Aziz HA, Hamdy NA, Farag AM, Fakhr IMI (2007) J Chin Chem Soc (in press)
- [19] Farag AM, Dawood KM, Abdel-Aziz HA (2004) J Chem Res 808
- [20] Dawood KM, Farag AM, Abdel-Aziz HA (2005) J Chem Res 278
- [21] Dawood KM, Farag AM, Abdel-Aziz HA (2004) Heteroatom Chem 7: 621
- [22] Dawood KM, Farag AM, Abdel-Aziz HA (2007) Heteroatom Chem **3**: 294

- [23] Dawood KM, Farag AM, Abdel-Aziz HA (2006) J Chin Chem Soc 53: 873
- [24] Farag AM (1995) J Chem Res (S) 96
- [25] Farag AM, Dawood KM, Kandeel ZE (1996) Tetrahedron 52(23): 7893
- [26] Farag AM, Dawood KM, Kandeel ZE (1997) Tetrahedron 53(1): 161
- [27] El-Merabani MM, El-Aaser AA, Attia MA (1972)J Planta Medica 36: 150
- [28] D'Amico JJ, Campbell RH, Guinn EC (1964) J Org Chem 19: 865
- [29] Duffin GF, Kendall JD (1956) J Chem Soc 361
- [30] Dieckmann W, Platz O (1905) Chem Ber 38: 2989
- [31] Hegarty F, Cashoman MP, Scoti FL (1971) Chem Commun 13: 884
- [32] Shawali AS, Osman A (1971) Tetrahedron 27: 2517